

mahajan and rangwala electricity pdf

In atomic physics, the Bohr magneton (symbol μ_B) is a physical constant and the natural unit for expressing the magnetic moment of an electron caused by either its orbital or spin angular momentum.. The Bohr magneton is defined in SI units by $\mu_B = \frac{e\hbar}{2m_e}$ and in Gaussian CGS units by $\mu_B = \frac{e\hbar}{2m_e c}$ where e is the elementary charge, \hbar is the reduced Planck constant, m_e is the electron rest mass and

Bohr magneton - Wikipedia

In atomic physics, the electron magnetic moment, or more specifically the electron magnetic dipole moment, is the magnetic moment of an electron caused by its intrinsic properties of spin and electric charge. The value of the electron magnetic moment is approximately 9.284764×10^{-24} J/T. The electron magnetic moment has been measured to an accuracy of 7.6 parts in 10^{13}

Electron magnetic moment - Wikipedia

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